Table 3. Selected geometric parameters (Å, °) for (2)

C9—H9 C9—C8a C9—C9a C9—C1'	0.95 (2) 1.530 (3) 1.525 (3) 1.528 (3)	C2'—C1' C2'—C7' C2'—C3'	1.407 (3) 1.548 (3) 1.402 (3)
H9—C9—C8a H9—C9—C9a H9—C9—C1'	107.7 (14) 108.0 (15) 109.1 (14)	C1'—C2'—C7' C3'—C2'—C7'	123.7 (2) 119.9 (2)
C1—C9a—C9—C1' C1—C9a—C9—H9 C4'—C3'—C2'—C7' C6'—C1'—C2'—C7'	-50.3 (3) 71.3 (14) -178.4 (2) 178.8 (2)	C1—C9a—C9—C8a C8—C8a—C9—C9a H9—C9—C1'—C2' H9—C9—C1'—C6'	-175.5 (2) 176.8 (2) 3.3 (14) -176.1 (14)

In both structures, the H9 atoms were refined isotropically. All other H atoms were riding.

For both compounds, data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1996); cell refinement: MSC/AFC Diffractometer Control Software; data reduction: TEXSAN PROCESS (Molecular Structure Corporation, 1995); program(s) used to solve structures: TEXSAN SHELXS86 (Sheldrick, 1985); program(s) used to refine structures: TEXSAN LS and SHELXL93 (Sheldrick, 1993); molecular graphics: TEXSAN ORTEP (Johnson, 1965); software used to prepare material for publication: TEXSAN, SHELXL93 and PLATON (Spek, 1990).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1380). Services for accessing these data are described at the back of the journal. A displacement ellipsoid plot of molecule (1b) has also been deposited.

#### References

Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.

Meyers, C. Y., Chan-Yu-King, R., Wahner, A. P., Manohar, S. K., Carr, S. E. & Robinson, P. D. (1991). Acta Cryst. C47, 1236–1239.
Meyers, C. Y., Hou, Y., Lutfi, H. G. & Robinson, P. D. (1995). Am. Chem. Soc. Natl Meet., Chicago, August 20–24, Abstracts ORGN 207

Meyers, C. Y., Hou, Y., Lutfi, H. G., Robinson, P. D., Dunn, H. E. & Seyler, J. W. (1997). Am. Chem Soc. Natl Meet., San Francisco, April 13-17, Abstracts ORGN 351.

Meyers, C. Y., Hou, Y., Scott, D. & Robinson, P. D. (1997). Acta Cryst. C53, 1149–1151.

Meyers, C. Y., Hou, Y., Scott, D., Robinson, P. D., Dunn, H. E. & Seyler, J. W. (1997). Am. Chem. Soc. Natl Meet., San Francisco, April 13–17, Abstracts ORGN 352.

Meyers, C. Y., Tunnell, J. L., Robinson, P. D., Hua, D. H. & Saha, S. (1992). Acta Cryst. C48, 1815–1818.

Molecular Structure Corporation (1995). TEXSAN. TEXRAY Structure Analysis Package. MSC, 3200 Research Forest Drive, The Woodlands. TX 77381, USA.

Molecular Structure Corporation (1996). MSC/AFC Diffractometer Control Software. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Nakamura, M., Nakamura, N. & Ōki, M. (1977a). Bull. Chem. Soc. Jpn, 50, 1097-1101.

Nakamura, M., Nakamura, N. & Ōki, M. (1977b). Bull. Chem. Soc. Jpn, 50, 2986-2990.

Öki, M. (1993). In *The Chemistry of Rotational Isomers*, Vol. 30. *Reactivity and Structure Concepts in Organic Chemistry*. New York: Springer-Verlag.

Robinson, P. D., Lutfi, H. G., Lim, L. W. & Meyers, C. Y. (1994). Acta Cryst. C50, 1728–1732.

Sheldrick, G. M. (1985). SHELXS86. Program for the Solution of Crystal Structures. University of Göttingen, Germany.

Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. University of Göttingen, Germany. Spek, A. L. (1990). Acta Cryst. A46, C-34.

Acta Cryst. (1998). C54, 77-79

## First Determination of the Absolute Configuration of an Atropisomeric Flavin Derivative

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#### Abstract

The crystal structure and absolute configuration of the (-)-enantiomer of 3-(4,6-dibromo-2-methylphenyl)-10-(4-tert-butylphenyl)pyrimido[4,5-b]quinoline-2,4(3H,10H)-dione methanol solvate, C<sub>28</sub>H<sub>23</sub>Br<sub>2</sub>N<sub>3</sub>O<sub>2</sub>.CH<sub>4</sub>O, have been determined. The absolute configuration is S. The asymmetric unit contains two crystallographically independent molecules which are related by a pseudo-inversion center.

#### **Comment**

In the course of studies to determine precisely the reaction mechanism of flavoenzyme (Walsh, 1979), various optically active 5-deazaflavin derivatives have been synthesized and their stereochemical reactivities have been investigated in detail (Tanaka *et al.*, 1987; Shinkai, Kawase *et al.*, 1989; Shinkai, Yamaguchi *et al.*, 1989; Kawamoto *et al.*, 1989, 1990, 1992, 1992*a,b*, 1994; Ohno *et al.*, 1994, 1996).

However, few determinations of the absolute configurations of these chiral flavoenzyme models have been achieved so far. Therefore, we synthesized the title flavin derivative, (I), and performed an X-ray crystallographic analysis of the (–)-enantiomer using the anomalous dispersion effect of the Br atoms. The asymmetric unit contains two molecules and corresponding bond lengths and angles do not differ significantly between these molecules.

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The pyrimidoquinoline rings are nearly planar, *i.e.* to within 0.126 (4) Å in molecule A and 0.134 Å in molecule B. The compound has an axial chirality with respect to the orientation of the dibromo-2-methylphenyl ring. The principal differences between the two molecules are in the dihedral angles between the aromatic rings. The dihedral angle between the pyrimidoquinoline and dibromo-2-methylphenyl rings are 113.2 and  $66.4^{\circ}$  in molecules A and B, respectively, and those between the pyrimidoquinoline and p-tert-butylphenyl rings are 96.6 and  $98.5^{\circ}$  in molecules A and B, respectively.

Full-matrix least-squares refinements using anomalous dispersion factors for all non-H atoms resulted in an R factor of 0.034 for the present structure and 0.054 for the other enantiomorph. Consequently the absolute configuration of (-)-(I) has been assigned to be S. The Flack parameter (Flack, 1983) also confirmed this assignment. There is a hydrogen bond between N1 of (I) and the hydroxyl group (O3—H24) of the methanol of crystallization.

This determination of the absolute configuration of chiral (I) is useful for future studies in bioorganic chem-

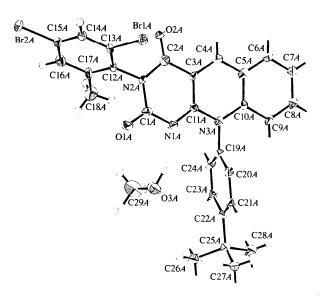


Fig. 1. *ORTEP*II (Johnson, 1976) drawing of molecule *A* of (I) showing displacement ellipsoids at the 50% probability level.

istry because it allows elucidation of the absolute configuration of the analog on debromination of compound (I) (Ohno *et al.*, 1996) and the CD spectrum of the debrominated compound can be used as a reliable basis for the conformational assignments of all 5-deazaflavin derivatives reported previously (Kawamoto *et al.*, 1992*a,b*).

### **Experimental**

The synthesis of (I) was carried out as reported previously (Ohno *et al.*, 1994) starting from 4,6-dibromo-*o*-toluidine, which was prepared by bromination of *o*-toluidine. The optical resolution was accomplished by HPLC [column, Chiralcel OD; eluent, ethanol; flow rate, 1.5 ml min<sup>-1</sup>; detection, UV 254 nm; retention times, 89.5 min for (+)- and 95.6 min for (-)-enantiomer] and both enantiomers were obtained in >99% e.e. The crystal of the (-)-enantiomer which was subjected to X-ray crystallographic analysis was obtained by recrystallization from methanol at room temperature.

#### Crystal data

C <sub>28</sub> H <sub>23</sub> Br <sub>2</sub> N <sub>3</sub> O <sub>2</sub> .CH <sub>4</sub> O	Mo $K\alpha$ radiation
$M_r = 625.36$	$\lambda = 0.71069 \text{ Å}$
Monoclinic	Cell parameters from 24
P2 <sub>1</sub>	reflections
a = 20.913 (7)  Å	$\theta = 14.9 - 15.1^{\circ}$
b = 10.868 (8)  Å	$\mu = 3.061 \text{ mm}^{-1}$
c = 12.03(1)  Å	T = 173.2  K
$\beta = 100.45 (4)^{\circ}$	Prismatic
$V = 2687 (2) \text{ Å}^3$	$0.4 \times 0.4 \times 0.2 \text{ mm}$
Z = 4	Yellow
$D_x = 1.545 \text{ Mg m}^{-3}$	
$D_m = 1.51 \text{ Mg m}^{-3}$	
$D_m$ measured by flotation in	
C <sub>6</sub> H <sub>14</sub> /CCl <sub>4</sub>	

#### Data collection

Rigaku AFC-7R diffractometer	8524 reflections with $I > 1.5\sigma(I)$
$\omega$ –2 $\theta$ scans	$R_{\rm int} = 0.045$
Absorption correction:	$\theta_{\text{max}} = 25.02^{\circ}$
$\psi$ scans (North, Phillips	$h = -24 \rightarrow 0$
& Mathews, 1968)	$k = -12 \rightarrow 12$
$T_{\min} = 0.440, T_{\max} = 0.542$	$l = -14 \rightarrow 14$
9752 measured reflections	3 standard reflections
9469 independent reflections	every 150 reflections
	intensity decay: 0.17%

#### Refinement

Refinement on F	Extinc
R = 0.034	Zacl
wR = 0.052	2 G
S = 0.862	Extinc
8524 reflections	0.00
667 parameters	Scatter
H atoms not refined	Inte
$w = 1/[\sigma^2(F_o) + 0.0025 F_o ^2]$	Cry:
$(\Delta/\sigma)_{\rm max} = 0.0204$	Absolu
$\Delta \rho_{\text{max}} = 0.43 \text{ e Å}^{-3}$	Flac
$\Delta \rho_{\min} = -0.70 \text{ e Å}^{-3}$	Flack

Extinction correction:
Zachariasen (1967) type
2 Gaussian isotropic
Extinction coefficient:
0.0002 (5)
Scattering factors from
International Tables for
Crystallography (Vol. C)
Absolute configuration:
Flack (1983)
Flack parameter = 0.021 (1)

Table 1. Selected geometric parameters (Å, °)

	_	_	
NIA—CIA	1.378 (7)	N3A—C11A	1.382 (6)
N1AC11A	1.314 (6)	N3A—C19A	1.451 (6)
N2A—C1A	1.435 (6)	C2A—C3A	1.489 (7)
N2A—C2A	1.379 (6)	C3A—C4A	1.368 (7)
N2A—C12A	1.445 (6)	C3A—C11A	1.429 (6)
N3A—C10A	1.389 (6)	C4A—C5A	1.418 (7)
C1A-N1A-C11A	118.4 (4)	C2A—C3A—C11A	117.6 (4)
C1A—N2A—C2A	123.9 (4)	C3A—C4A—C5A	120.6 (4)
C1A—N2A—C12A	118.4 (4)	C4AC5AC10A	117.7 (4)
C10A—N3A—C11A	122.5 (4)	N3A—C10A—C5A	120.4 (4)
C11A—N3A—C19A	119.2 (4)	N1AC11AN3A	118.1 (4)
N1A—C1A—N2A	119.5 (4)	N1AC11AC3A	125.2 (4)
N2A—C2A—C3A	114.2 (4)	N3A—C11A—C3A	116.7 (4)
C2A—C3A—C4A	120.3 (4)		

The space group was uniquely determined from the systematic absence: 0k0 when k = 2n + 1. Bijvoet pairs were not averaged. The non-H atoms were refined anisotropically and H atoms were placed in calculated positions but not refined.

Data collection: MSCIAFC Diffractometer Control Software (Molecular Structure Corporation, 1992). Cell refinement: MSCIAFC Diffractometer Control Software. Data reduction: TEXSAN PROCESS (Molecular Structure Corporation, 1995). Program(s) used to solve structure: direct methods SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: TEXSAN LS. Software used to prepare material for publication: TEXSAN FINISH.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: OA1036). Services for accessing these data are described at the back of the journal.

### References

Flack, H. D. (1983). Acta Cryst. A39, 876-881.

Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.

Kawamoto, T., Taga, T., Bessho, K., Yoneda, F. & Hayami, J. (1994). Tetrahedron Lett. 35, 8631-8634.

Kawamoto, T., Tanaka, K., Kuroda, Y. & Yoneda, F. (1990). Chem. Lett. pp. 1197-1200.

Kawamoto, T., Tanaka, K., Yoneda, F. & Hayami, J. (1989). Tetrahedron Lett. 30, 7431-7434.

Kawamoto, T., Tomishima, M., Kunitomo, J., Yoneda, F. & Hayami, J. (1992). *Tetrahedron Lett.* 33, 7173-7176.

Kawamoto, T., Tomishima, M., Yoneda, F. & Hayami, J. (1992a). Tetrahedron Lett. 33, 3169-3172.

Kawamoto, T., Tomishima, M., Yoneda, F. & Hayami, J. (1992b). Tetrahedron Lett. 33, 3173-3176.

Molecular Structure Corporation (1992). MSC/AFC Diffractometer Control Software. Version 4.3.0. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Molecular Structure Corporation (1995). TEXSAN. TEXRAY Structure Analysis Package. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351-359. Ohno, A., Kunitomo, J., Kawai, Y., Kawamoto, T., Tomishima, M., Bessho, K. & Yoneda, F. (1996). J. Org. Chem. 61, 9344-9355.

Ohno, A., Kunitomo, J., Kawamoto, T., Tomishima, M., Bessho, K. & Yoneda, F. (1994). *Tetrahedron Lett.* 35, 9729-9732.

Sheldrick, G. M. (1985). SHELXS86. Program for the Solution of Crystal Structures. University of Göttingen, Germany.

Shinkai, S., Kawase, A., Yamaguchi, T., Manabe, O., Wada, Y., Yoneda, F., Ohta, Y. & Nishimoto, K. (1989). J. Am. Chem. Soc. 111, 4928-4935, and references therein.

Shinkai, S., Yamaguchi, T., Kawase, A., Manabe, O. & Kellogg, R. M. (1989). J. Am. Chem. Soc. 111, 4935–4940, and references therein. Tanaka, K., Kimura, T., Okada, T., Chen, X. & Yoneda, F. (1987). Chem. Pharm. Bull. 35, 1397–1404.

Walsh, C. (1979). Enzymatic Reaction Mechanism, pp. 358–431. San Francisco: W. H. Freeman.

Zachariasen, W. H. (1967). Acta Cryst. 23, 558-564.

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# 3-[(Z)-Piperidin-1-ylmethylidene]-2,3-dihydro-1,4-benzodioxan-2-one

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#### **Abstract**

The crystal structure of the title compound has been determined in order to obtain the geometry of one isomer of  $C_{14}H_{15}NO_3$ . The molecule has the unusual feature of a planar dioxane ring and several bond angles are enlarged due to steric hindrance.

#### Comment

In connection with our investigations into 1,4-benzo-dioxane chemistry, the reaction between 1,4-benzo-dioxane-2-carboxylic acid and some amines has been studied (Ruiz et al., 1996), and the title compound, (I), was isolated. Although spectroscopic data show that (I) is clearly different from its isomer 2-piperidinyl-carbonyl-1,4-benzodioxane, X-ray diffraction analysis has permitted the determination of the structure and geometry of this new unexpected compound.

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